AMENDMENTS TO THE SPECIFICATION

The paragraph beginning at page 9, line 10, has been amended to read as follows:

Advantageously, the hydrogenated inorganic compound is chosen from the group

composed of alumina hydrates and magnesium hydroxide (Mg(OH)₂), while the inorganic boron

compound is chosen from the group composed of boric acid (H₃BO₃), colemanite

 $(Ca_2O_{14}B_6H_{10})$, zinc borates $(Zn_2O_{14.5}H_7B_6, Zn_2O_{14.5}H_7B_6, Zn_4O_8B_2H_2, Zn_2O_{11}B_6)$, boron

carbide (B₄C), boron nitride (BN) and boron oxide (B₂O₃).

The paragraph beginning at page 12, line 25, has been amended to read as follows:

After degassing, the mix is poured into the required mould where it sets, due to

polymerization of the vinylester resin, and is transformed into an insoluble material. This is a

radicalar radical type polymerization and it is highly exothermal. The setting time may vary

depending on pour conditions (temperature, catalyst, accelerator content, etc.). Thus, the gel

time may be varied by varying the percentages of catalyst and accelerator. The gel time varies

from 20 minutes to 2 hours.

The paragraph beginning at page 15, line 14, has been amended to read as follows:

Table 1 also shows the density, hydrogen and boron contents, the vitreous transition

temperature (Tg), the coefficient of thermal expansion (a), the specific heat (Cp) and the thermal

conductivity (\(\lambda\)) of a material hereinafter referred to as the "reference material" formed after

setting of a mix of the following constituents (25 minutes at 20°C), for comparison purposes:

• 32% by mass of this mix being the novolac-type vinylester resin of the Derakane

Momentum® 470-300,

• 62% by mass of this mix being the alumina hydrate SH 150/01, and

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• 6% by mass of this mix being a zinc borate Zn₂O_{14,5}H₇B₆ Zn₂O_{14.5}H₇B₆ (Firebrake ZB - BORAX),

plus the accelerator NL 49P (0.9% by mass of the resin) and the catalyst Butanox® M50 (1.5% by mass of the resin).